

Lithium Ion & Lithium Polymer Batteries

Daren Slee, P.E., CRE Exponent, Inc.

A leading engineering & scientific consulting firm dedicated to helping our clients solve their technical problems.



Who We Are

Exponent is a multi-disciplinary consulting firm dedicated to solving important science, engineering and regulatory issues for clients





Exponent Offices







Engineering Sciences

- Biomedical Engineering
- Electrical & Semiconductors
- Engineering Management Consulting
- Mechanical Engineering
- Materials & Corrosion Engineering
- Polymer Science & Materials Chemistry
- Thermal Sciences

Environmental **Sciences**

- Ecological & Biological Sciences
- Environmental & Earth Sciences







Health Sciences

- Chemical Registration & Food Safetv
- Epidemiology, Biostatistics & **Computational Biology**
- Exposure Assessment & **Dose Reconstruction**
- Occupational Medicine & Environmental Health
- Toxicology & Mechanistic Biology

Transportation

- Biomechanics
- Human Factors
- Statistical & Data Sciences
- Vehicle Engineering
- Visual Communications

Civil & Construction

- Buildings & Structures
- Civil Engineering
- Construction Consulting
- Industrial Structures















General Battery Hazards

- High Instantaneous Current
- Voltage Dependent on Number of Cells in Series
 - High Voltage Can Result in an Arc Explosion
 - High Voltage and Current Equals High Power
 - High Power Results in Explosive Energy Release
 - Similar to HV AC Source
 - Injuries and Death Can Occur
 - Possible Fire Ignition

High Voltage Can Result in Shock Injuries or Electrocution

- Large Amount of Energy Available to a Load Fault
 - Batteries Usually Fused to Prevent Large Fault Current
 - Example: Automotive Fuses
 - However, If Protection Fails Fire Can Occur



Battery Energy Release

- Flammable battery electrolyte (Lithium Ion)
- Why are they used?
 - Lithium Ion Energy Density
 - ~150 Wh/kg
 - ~200 Wh/L
 - Nickel Metal Hydride Energy Density
 - ~100 Wh/kg
 - ~100 Wh/L
 - Nickel Cadmium Energy Density
 - ~60 Wh/kg
 - ~70 Wh/L
 - Lead Acid Energy Density
 - ~40 Wh/kg
 - ~65 Wh/L

Nominal Cell Voltage: 3.6V

Nominal Cell Voltage: 1.2V

Nominal Cell Voltage: 1.2V

Nominal Cell Voltage: 2V



Challenges

- Lithium Ion batteries are significantly different in every aspect compared to traditional battery chemistries
 - Organic electrolyte
 - Strong oxidizers and reducers
 - No recombination rate ability

... requires failsafe controls

- Cell is manufactured at one location ...battery at another, product at another...
 - ... yet all needs to fit and work together



Lithium Battery Powered Systems

• What are the main issues?

- Chemistry
- Electrical system Arcing/Shock and Electrocution
- Manufacturing
- Recalls
- Accidents
- Pack integration architecture Module separation choice of insulator solutions
- Protection circuit and redundancy in protection systems



The Pack, Host Device and Accessories

- Critical sub-systems responsible for maintaining a suitable environment for the cells
 - Mechanical protection
 - FMEA
 - "Real World" mechanical testing
 - Environmental protection
 - Use profile temperature cycling
 - Cycling with exposure to expected (or unexpected) conditions
 - Electrical stability within operational windows
 - Safety's and limits maintained over all use and foreseeable misuse conditions



Focus needs to be on the electrical and electronic design too

- Attention to circuit design and layout copper traces
- Sufficient protection in the design
- Independent safety protection redundancy
- Choice of components
- Effects of high voltage
- Connectivity
- Characterization of the worst case scenarios multiple points of failure



Lithium Ion Basics

Positive Electrode (thicker white spiral in scan)

- Aluminum Current Collector
- Coated with LiCoO₂ Active material

Negative Electrode (thinner...)

- Copper Current Collector
- Coated with Graphite

Electrolyte

- Ethylene Carbonate
- LiPF₆ Salt

"Jellyroll" is wrapped electrodes with electrolyte injected

CT Cross-section scan



Lithium Ion Basics

- Copper is used as the negative electrode because if aluminum is used the aluminum participates as an ion in the charge and discharge reactions causing corrosion
- Reaction Equation:

Charged Discharged Li_{1/2}CoO₂ + Li_{1/2}C₆ \leftrightarrow C₆ + LiCoO₂

Types of Lithium Ion Batteries

- Cylindrical cells use nickel-plated steel cans
 The cell can is at the cell negative potential
- Prismatic cells typically use aluminum cans
 - The cell can is at the cell positive potential
 - Some larger, heavier prismatic cells use nickel-plated steel cans (can at cell negative potential)

Polymer cells use a polymer coated aluminum foil pouch

- Pouch is left electrically floating and is insulated from both the positive and negative terminals of the cell
- More sensitive to mechanical abuse

Lithium Ion Battery Failure Analysis Use a Fault Tree Analysis (FTA) approach

- The root causes discussed are the branches of the tree
- Cut off branches that are not consistent with the evidence
- Remaining branches evaluated to rank relative likelihood as root cause
- Test electronics to determine functionality of charge and protection circuits
- Analyze heat and mechanical damage patterns to determine if they are external
- Analyze the damage to the protection circuit for evidence of an electronics failure
- Most often "internal cell fault" is the only branch remaining

Manufacturing Issues

- Microparticle contamination in the active material slurries used to coat the electrodes
- Assembly line tools wear and can drop particles into the cell raw materials
- Cutter blades dull leaving burrs and tears on the current collector metals and leads connected to the foils
- Rough lead to foil connection techniques can leave sharp edges
- Nickel plating on substandard cell cans and other construction materials can flake and drop into the "jellyroll" during cell construction

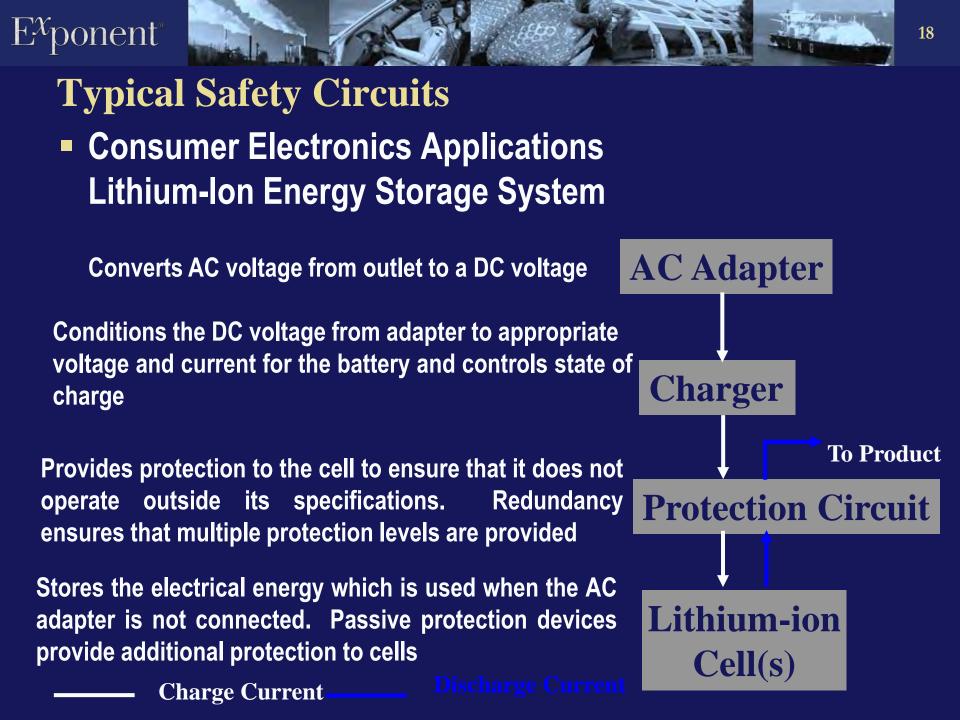


Take Control While You Can

- The Cells
 - Forget the spec sheet test to device requirements
 - Confirm quality and continue to check
 - Shop for a deal...but don't get burned
- The Pack, Host Device and Accessories
 - Don't stop with the standards and guidelines understand the possible failure modes and design away from them
 - Simple, robust circuits based on accepted designs
 - Redundancy for critical operations
 - Respect for the limits of the cells
 - Mechanical integrity sufficient for the intended use and foreseeable environmental conditions

The Cells: Qualify, Confirm and Check

- Qualify cells for the intended application
 - Test to the specifications of the device
 - Test under normal use conditions
 - Test under reasonable abuse conditions
- Confirm you are getting what you paid for
 - Assess and record build quality and workmanship
 - Consider analytical work for custom designs
- Check your incoming material on a regular basis
 - Establish an incoming QC procedure
 - Catch problems on the inside



Overcharge Protection

Multiple Cell Application: 4 Independent Levels

- 1. Charger Output Voltage
- 2. Battery Protective Switch
- 3. Electronically Controlled Fuse
- 4. 18650 Cell Current Interrupt Device (CID)
 - Prismatic and Polymer Designs Use Thermal Cutoffs (TCO)

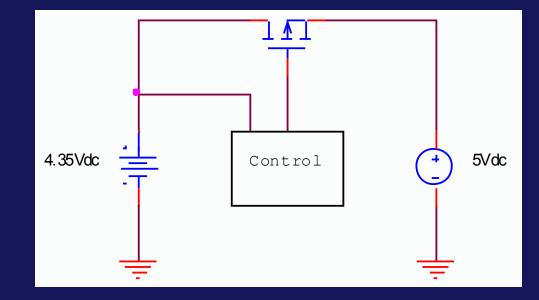
Single Cell Application: 2 Independent Levels

- 1. Charger Output Voltage
- 2. Battery Protective Switch



Overcharge Protection

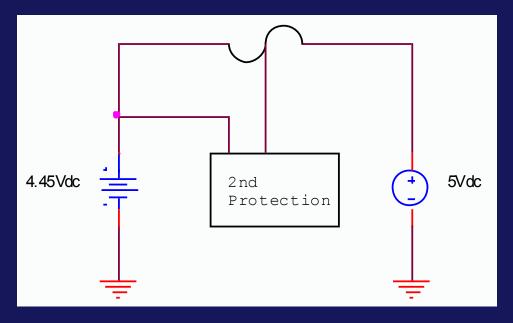
- Battery Protective Switch
- Microcontroller controlled transistor switch
- In multiple cell applications each cell is individually monitored for overcharge by the microcontroller





Overcharge Protection

- Independent IC: Secondary Protection
- Each cell is individually monitored for overcharge
- Electronically controlled fuse is opened if overcharge is detected



Overcurrent Protection

Multiple Cell Application:

- 4 Independent Levels for Charge Current
- 3 Independent Levels for Discharge Current
 - 1. Charger Current Limit (Charger Only)
 - 2. Battery Protective Switch
 - 3. Standard Current Fuse
 - 4. Positive Temperature Coefficient Device (PTC)
 - Prismatic and Polymer Designs Integrate PTCs external to the cells

Single Cell Application:

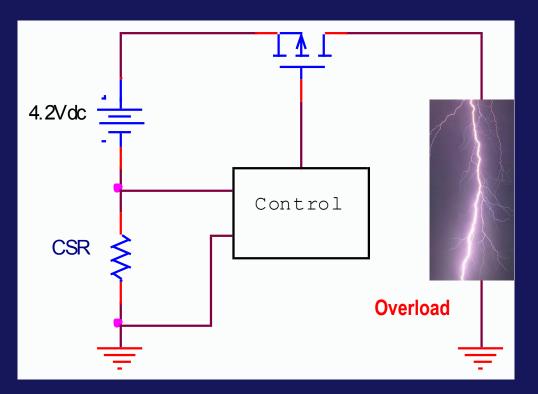
- 3 Independent Levels for Charge Current
- 2 Independent Levels for Discharge Current
 - 1. Charger Current Limit (Charger Only)
 - 2. Battery Protective Switch
 - 3. PTC external to cell



Overcurrent Protection

Battery Protective Switch

- Microcontroller monitors charge and discharge current using a Current Sense Resistor (CSR)
 - Or FET On-Resistance
- If overcurrent detected
 - Protective Switch Opens

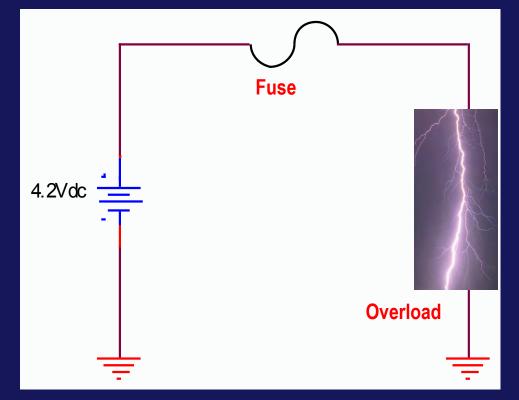




Overcurrent Protection

Standard Current Fuse

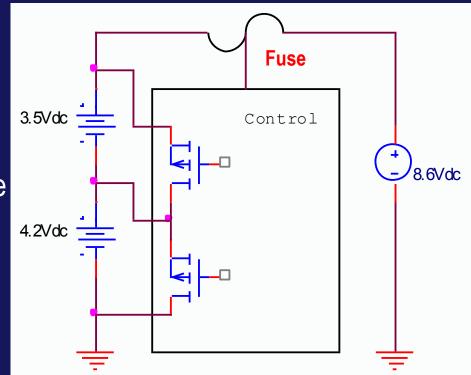
 Overcurrent will cause the fuse to open



Imbalance Protection

Series connected cells require imbalance protection

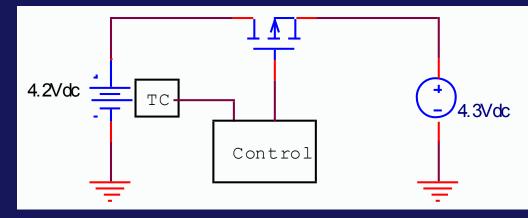
- Microcontroller monitors individual cell voltages
- Methods of protection include:
 - Rebalancing by diverting charge current from higher voltage cells
 - If imbalance is severe, electronically controlled fuse can be opened





Overtemperature Protection

- Microcontroller senses cell temperature using a thermocouple (TC) and terminates charge or discharge current using a switch based on the following typical criteria:
 - Charge initiation allowed within 0°C and 50°C range
 - Charge continuation allowed within 0°C and 60°C range
 - Discharge allowed within 0°C and 70°C range

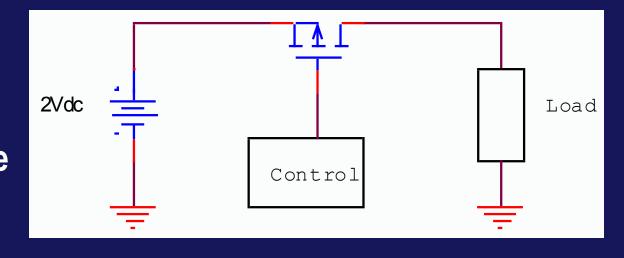


Overdischarge Protection

Microcontroller monitors cell voltages

- Opens switch when the capacity of the battery is drained
- Drained battery will be drained further by protection electronic loading
- Protection electronics will shut down when the cell voltages drop below 2V

 Drained pack can be reenabled with low "pre-charge" current to bring the cells back into normal range



Large Battery Systems

- Cell designs that assist in the distribution of heat
- Ceramic separators to improve thermal stability
- Positive electrode material with greater thermal stability
- Case designs to improve heat transfer (fins etc.)
- Forced convection mechanisms for heat transfer
- Soft packages for cells to provide larger aspect ratios to aid in better heat transfer

Large Battery Systems

Electrical Shock Hazard

- HEV Li-ion battery systems have substantially higher operating voltages
- Typically 160 V or higher
- UL defines a voltage in excess of 42.4 Vac or 60 Vdc as hazardous

Arcing

- An arcing fault can result in extremely high temperatures on the order of 10,000° C or higher.
- These high temperatures can generate hot gases and molten metal which can result in serious burns and cause clothing to ignite



Test standards Lithium-Ion Abuse Standards

Underwriters Laboratories

- UL 1642: Lithium Batteries
- UL 1973 (Proposed): Batteries for use in Light Electric Rail (LER) Applications and Stationary Applications
- UL 2054: Household and Commercial Batteries
- UL 2271: Batteries for use in light electric vehicle applications
- UL 2580: Batteries for use in electric vehicles

Institute of Electrical and Electronics Engineers (IEEE)

- IEEE 1625: Rechargeable Batteries for Multi-Cell Mobile Computing Devices
- IEEE 1725: Rechargeable Batteries for Cellular Telephones
- American National Standard (ANSI)
 - C18.2M Part2: Portable Rechargeable Cells and Batteries Safety Standard
- Society of Automotive Engineers (SAE)
 - J2464: Electric and Hybrid Electric Vehicle Rechargeable Energy Storage Systems (RESS), Safety and Abuse Testing
 - J2929: Electric and hybrid vehicle propulsion battery system safety standard lithium based rechargeable cells



Lithium-Ion Abuse Standards

International Electrotechnical Commission

- IEC 62133: Secondary cells and batteries containing alkaline or other non-acid electrolytes - Safety requirements for portable sealed secondary cells, and for batteries made from them, for use in portable applications
- IEC 62281: Safety of Primary and Secondary Lithium cells and batteries during transport

United Nations (UN)

 Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria, Part III, Section 38.3

Japanese Standards Association

 JIS C8714: Safety tests for portable Lithium-ion secondary cells and batteries for use in portable electronic applications

Battery Safety Organization (BSO)

 Proposed: Manual for the evaluation of energy systems for light electric vehicle (LEV) – secondary lithium batteries.



Battery Systems Support Overview

- Failure analysis and root cause determination
 - Litigation
 - Field Incidents
 - Recall support
- Design evaluation and review
- Standard and customized testing abuse/misuse testing
- Battery Accelerated aging and predictive life modeling
- System integration support

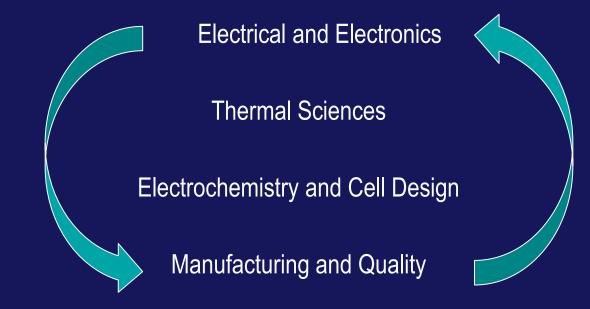
Vehicle \rightarrow Electrical System \rightarrow Battery Pack \rightarrow Electrochemical Cells



Multidisciplinary Approach

By forming a multidisciplinary team we have experience with:

- Cells
- Power management
- Electrical system
- Risk analysis
- Auditing of component factories



Equipment and Capabilities: Phoenix

- Pack, System and Vehicle Level Support:
 - circuit design and layout
 - protection and redundancy in the design
 - choice of components
 - effects of high voltage
 - environmental impact on performance (water, temperature, humidity, etc...)
 - connectivity
 - characterization and testing of the worst case scenarios
- Equipment and Facilities:
 - Environmental chambers (-25 °C to 100 °C)
 - Shock table
 - Vibration table
 - Thermal imaging camera
 - Vacuum chambers
 - High speed video capability (up to 2000 frames/second)
 - Maccor cell cyclers
 - Complete vehicle testing center



